

INTRODUCTION

- Pistoia Alliance and CAS: Not-for-profit organizations committed on obtaining crowd sourced data to promote safety in laboratories
- Pistoia's CSL: An open access platform developed by CAS that provides hazardous reaction information to alert scientists of dangerous reactions
- Goal: Expand the CSL by Identifying new hazardous 2+ reagent reactions
- Method: Develop model to find most reactive chemicals to enter into CSL database



EXAMPLE SUBMISSION / ENTRIES TO CSL

1	Reagent/Reactants Involved in Incident Report*	
Incldent	Reagent/Reactant 1 - Name	Reagent(s)
Report For	m Description of the Description Number	BROMINE
	Reagent/Reactant 1 - CAS Registry Number Warning Message* Please enter a brief description of what happened and what could have been done differently.	CAS RN: 7726-95-6 MFCD: MFCD00010896 SMILES: BrBr InChi: InChI=1S/Br2/c1-2 InChi Key: GDTBXPJZTBHREO-UHFFFAOYSA-N
Sample En	try	SULFURIC ACID CAS RN: 7664-93-9 MFCD: MFCD00064589 SMILES: OS(=O)(=O)O InChi: InChI=1S/H2O4S/c1-
Date Created: 11/5/17 Modified Date 5/24/18	Warning Message: Warning- Incorrect concentrations of sodium bromide with concentrated sulfuric acid	5(2,3)4/h(H2,1,2,3,4) InChi Key: QAOWNCQODCNURD- UHFFFAOYSA-N
	can cause bromine gas to be released GHS Category: Corrosive, Toxic Functional Group: - Reaction Class: CSL00040 Bromination BROMINE Source:	SODIUM BROMIDE CAS RN: 59217-63-9, 7647-15-6 MFCD: MFCD00003475 SMILES: [Na+].[Br-] InChi: InChI=1S/BrH.Na/h1H;/q;+1/p-1 InChi Key: JHJLBTNAGRQEKS-UHFFFAOYSA-M
	User-Reported	

Chemical Safety Library

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9.2 Reactivity Profile

Mixing acetic acid in equal molar portions with any of the following substances in a closed container caused the temperature and pressure to increase; 2-Aminoethanol, chlorosulfonic acid, ethylene diamine, ethyleneimine [NFPA 1991]. Acetic acid or acetic anhydride can explode with nitric acid if not kept cold. Potassium hydroxide residue in a catalyst pot reacted violently when acetic acid was added [MCA Case History 920, 1963]. During the production of terephthalic acid, n-xylene is oxidized in the presence of acetic acid. During these processes, detonating mixtures may be produced. Addition of a small amount of water may largely eliminate the risk of explosion [NFPA 491M.1991.p. 7]. Acetaldehyde was put in drums previously pickled with acetic acid. The acid caused the acetaldehyde to polymerize and the drums got hot and vented [MCA Case History 1764. 1971]. A mixture of ammonium nitrate and acetic acid ignites when warmed, especially if concentrated [Von Schwartz 1918. p. 322]. Several laboratory explosions have been reported using acetic acid and phosphorus trichloride to form acetyl chloride. Poor heat control probably caused the formation of phosphine [J. Am. Chem. Soc. 60:488. 1938]. Acetic acid forms explosive mixtures with p-xylene and air (Shraer, B.I. 1970. Khim. Prom. 46(10):747-750.).

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METHODOLOGY 2: REACTIVITY PROFILE EXTRACTION

Web-scraped PubChem using the PUG-VIEW API to request information such as the reactivity profile and IUPAC name for a particular compound
Split reactivity profile into sentences and extracted compounds, shown as embedded links, to form reactant groups along with the specific compound
Filtered out sentences not containing hazard words to create warning messages for the reactant group

Reactant 1: hydrogen

Reactant 2: 1-aminopropan-2-ol Warning Message: Flammable gaseous hydrogen is generated by amines in combination with strong reducing agents, such as hydrides.

Reactant 1: hydrazine hydrate Reactant 2: 1-chloro-2,4-dinitrobenzene

Warning Message: The mixture of this compound with hydrazine hydrate caused a violent reaction.

Reactant 1: ammonia Reactant 2: 1.2-dichloroethane

Warning Message: Liquid ammonia and ethylene dichloride can cause an explosion when mixed, nfpa 491m, 1991.

• Tf-idf model to sort reactivity profiles by highest frequency of

hazard words

- Compares individual reactivity
 profile to pre-determined list of
 words
- Determines most dangerous
 compounds and reactions to
 add to CSL first

Chemical Name	Similarity Score
Acetyl Chloride	0.62683517
Isopropylamine	0.6027288
Bromoform	0.55904955
Ethyl Chloride	0.5437039

CONCLUSION

Successfully built up models to extract reactivity information from PubChem Sorted through the corpus using dangerous word frequencies

Formed an organized list of 2+ reagent reactions with warning messages that can be manually inputted into the CSL

Next steps:

- $\circ~$ Running key word searches on different data sources
- $\circ~$ Automation of incident report filing
- $\circ~$ Include reactions of 3+ reagents

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References:							
<u>PubChem</u>	<u>Gensim</u>	<u>PUG</u>	Cameo Chemicals				